

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 3429-3432

Relative stereochemistries of the ether rings and sugar moieties in durinskiol A

Masaki Kita,^a Michael C. Roy,^{b,†} Eric R. O. Siwu,^b Isao Noma,^b Takahiro Takiguchi,^b Kaoru Yamada,^b Tomoyuki Koyama,^{b,‡} Takeshi Iwashita,^c Atsushi Wakamiya^b and Daisuke Uemura^{b,*}

^aResearch Center for Materials Science, Nagoya University, Furo-cho, Chikusa, Nagoya 464-8602, Japan ^bDepartment of Chemistry, Graduate School of Science, Nagoya University, Furo-cho, Chikusa, Nagoya 464-8602, Japan ^cSuntory Institute for Bioorganic Research, Mishima, Osaka 618-8503, Japan

> Received 15 February 2007; revised 5 March 2007; accepted 8 March 2007 Available online 12 March 2007

Abstract—Durinskiol A (1) is a long carbon-chain polyol compound with a molecular weight of 2128 mu isolated from the symbiotic dinoflagellate *Durinskia* sp. The relative stereochemistries of ether rings and sugar moieties in 1 were established based on 2D-NMR analysis and molecular modeling studies. Conformational analysis of the 6,5,6-bis-spiroacetal ring in 1 was also examined by using ab initio methods.

© 2007 Elsevier Ltd. All rights reserved.

In a previous paper, we reported isolation, planar structure, and biological activity of durinskiol A (1), a unique polyol compound with a molecular weight of 2128 mu.¹ The structural features of this molecule are a 6,5,6-bisspiroacetal ring, α - and ω -olefins on the termini of the C₉₃ linear carbon-chain, two sugar units, and six- and seven-membered ether rings. We report here the relative stereochemistry of durinskiol A (1) determined by spectroscopic analysis and molecular modeling studies (Fig. 1).

The relative configurations of the ether rings and sugar units in durinskiol A (1) were analyzed using NOESY, ROESY, NOEDF and homonuclear *J*-resolved NMR spectra (Fig. 2). Regarding the ABC-ring, key NOEs were observed for H17/H19, H20b/CH₃99, and H25/ H27a, which suggested that three ethereal oxygen atoms

0040-4039/\$ - see front matter @ 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2007.03.039

faced each other. The large magnitude of $J_{17,18b}$, $J_{27a,28}$, and $J_{28,29a}$ (9.6–10 Hz) suggested that the four protons H17, H19, H27a, and H28 may be oriented in axial orientations with respect to the ether rings A and C with a chair conformation due to the minimization of steric repulsions between equatorial substituents, and by anomeric effects in an axial C-O conformation in the spiro[4.5] system. A molecular modeling study using a Merck molecular force field (MMFF) for the ABC-ring model compound 2 also indicated that the same conformation as above was appropriate (Fig. 3), and the calculated distances of all protons for which NOEs were observed in rings ABC in 1 were less than 2.7 Å. To obtain more reliable conformational energies, geometry optimization for the possible four conformers of 2 was conducted using B3LYP/6-31G(d) level of theory,² whose ethereal oxygen atoms were oriented in axial/ equatorial (i), bis-axial (ii), bis-equatorial (iii), and equatorial/axial (iv), respectively (Fig. 4). Their having minimum energies were confirmed by frequency calculations. As a result, the axial/equatorial conformer (i) was found to be most stable, and conformers (ii), (iii), and (iv) were higher in energy by 3.8, 10.2, and 11.3 kcal/mol than conformer (i). Although conformer (ii) may incorporate the maximum amount of anomeric effects, 1,3-diaxial interactions between the axial substituent on C28 and the ethereal oxygen atom in the ring B

Keywords: Configuration determination; Glycosides; Molecular modeling; Natural products; Super-carbon-chain compounds.

^{*} Corresponding author. Tel./fax: +81 52 789 3654; e-mail: uemura@ chem3.chem.nagoya-u.ac.jp

[†]Present address: Okinawa Institute of Science and Technology, 7542 Onna, Onna-Son, Kunigami, Okinawa 904-0411, Japan.

[‡]Present address: Graduate School of Marine Science and Technology, Tokyo University of Marine Science and Technology, 4-5-7 Konan, Minato-ku, Tokyo 108-8477, Japan.



Figure 1. Structure of durinskiol A (1). The relative stereochemistry between rings C/D, E/F, G/xylose, and H/rhamnose has not been clarified.



Figure 2. Relative stereochemistry of rings ABC, DE, FGH, and two sugar moieties for 1. Asterisk indicates interchangeable values.

may be unfavorable.³ Thus, the relative stereochemistry in rings ABC was established to be $17S^*$, $19S^*$, $21R^*$, $24S^*$, $25R^*$, and $28S^{*.4}$

As for the D ring, the large magnitude of $J_{34a,35}$, $J_{35,36}$, $J_{36,37}$, $J_{37,38}$, and $J_{38,39a}$ (8.8–10.4 Hz) suggested that all of the five protons H34a, H35, H36, H37, and H38 were oriented in *anti* arrangements with respect to the D ring. Key NOEs H33/H35, H34a/H36, and H35/H37 indicated that the two alkyl substituents (C32 and C39) and three hydroxyl groups in the D ring should be oriented in pseudoequatorial configurations with respect to the seven-membered ether ring with a chair conformation. Based on the molecular modeling study for the D-ring model compound **3**, all of the observed NOEs were applicable as with **2** (Fig. 3). Furthermore, the *J* values of vicinal protons (H34–H38) estimated from their dihedral

angles in the calculated model compound **3** mostly coincided with those of natural **1** (Table 1). Thus, the D-ring was confirmed to be a trans-fused ether.

Based on the NOEs observed for H40/H45, H42/H44a, H50/H52/CH₃98, H60/H63a/CH₃97, and H64/H65/H67, the F ring was found to be *cis*-2,6-dialkyltetrahydropyran, and the other three were trans. The large magnitude of $J_{50,51a}$, $J_{51a,52}$, $J_{60,61a}$, $J_{66a,67}$, and $J_{67,68}$ (9.3–11.3 Hz) suggested that the seven protons H50, H52, H60, H65, H66a, H67, and H68 were oriented in axial configurations with respect to the F, G, and H rings. Similarly, axial orientations for the two hydroxyl groups of C43 and C53 were established by the small magnitude of $J_{H42,H43}$ (2.0 Hz) and $J_{H52,53}$ (3.6 Hz), respectively. Thus, relative stereochemistry of each ethereal E-H ring part was confirmed.



Figure 3. Three-dimensional structures of the ABC-ring model 2 and the D-ring model 3. Each model was energetically minimized via iterative application of the MMFF (MacroModel 9.0). Values represent the calculated distances between two selected protons (Å). Substituent groups on C16 and C29 in 2 and on C33 and C39 in 3 were replaced with methyl groups and hydroxyl methyl groups, respectively.



Figure 4. Optimized structures and relative energy of four conformers for 2 at the B3LYP/6-31G(d) level of theory.

Regarding the D/E linkage, the *anti* correlations for H38–C38–C39–H39a and H39b–C39–C40–H40 were established by the large magnitudes of $J_{\rm H38,H39a}$ (9.2 Hz) and $J_{\rm H39b,H40}$ (8.8 Hz). The obvious NOEs H37/H39b, H38/H39b, H39a/H41, and H40/45 strongly indicated that the hydroxyl group on C40 was oriented in a pseudoaxial configuration with a zigzag configuration from the carbon atom (C36) in D ring to the ethe-

Table 1. Comparison of ${}^{1}H{-}^{1}H$ coupling constants in the D-ring model compound 3 and natural 1^{5}

Coupling protons	Dihedral angles in 3 (deg)	Calculated coupling constants in 3 (Hz)	Observed coupling constants in 1 (Hz)
H33/H34a	-156.5	8.5	N.D.
H33/H34b	86.2	1.0	5.7
H34a/H35	179.3	11.2	10.4
H34b/H35	-65.6	3.5	3.7
H35/H36	163.5	6.4	8.8
H36/H37	179.7	9.4	9.2
H37/H38	-156.3	7.6	9.2

N.D.: Not determined.

real O atom in E ring, which alone satisfied all of the observed NOEs and J values. Thus, the relative stereochemistry in the DE rings was established to be $33R^*$, $35R^*$, $36S^*$, $37S^*$, $38S^*$, $40S^*$, $41S^*$, $42S^*$, $43S^*$, and 45S*. The strong NOEs observed for H53/H56/H59/ H1' suggested that these four protons face each other with a zigzag conformation from carbons C54 to C58 between the F/G rings. The methylene protons H63a (δ 1.25) and H63b (δ 2.30) were clearly distinguished with each other. Key NOEs for H60/H63a, H60/H66a, H62/H63b, and H63b/H64 indicated that the free rotation of two sp³-sp³ bonds, C62-C63 and C63-C64, was rather restricted and that only one configuration shown in Figure 2 could satisfy all of these data. Thus, the relative stereochemistry in the FGH rings was established to be $50S^*$, $52R^*$, $53R^*$, $54R^*$, $56S^*$, $58S^*$, $59S^*$, $60S^*$, 62R*, 64R*, 65R*, 67S*, and 68R*.

The large magnitude of $J_{1',2'}$, $J_{2',3'}$, $J_{3',4'}$, $J_{4',5'a}$, $J_{3'',4''}$, and $J_{4'',5''}$ (7.8–10.5 Hz) suggested that all of the five hydroxyl groups on C2', C3', C4', C3'', and C4'' and a methyl group (C6'') were oriented in equatorial configurations. Meanwhile, an axial orientation for the hydroxyl group on C2'' was suggested by the small magnitudes of $J_{1'',2''}$ (1.2 Hz) and $J_{2'',3''}$ (2.8 Hz). Thus, two sugar moieties in **1** were unequivocally confirmed to be xylose and rhamnose.

In summary, a novel polyol compound with a molecular weight of 2128 mu, durinskiol A (1), was isolated from the dinoflagellate *Durinskia* sp., and its partial relative stereostructure was elucidated. Further chemical degradation studies and stereochemical assignment including the absolute stereochemistry of 1 are in progress.

Acknowledgments

This work was supported in part by a Grant-in-Aid for Creative Scientific Research (No. 16GS0206 for D.U.) from JSPS. We are also indebted to Ono Pharmaceutical Co., Ltd for their financial support.

References and notes

 Kita, M.; Roy, M. C.; Siwu, E. R. O.; Noma, I.; Takiguchi, T.; Itoh, M.; Yamada, K.; Koyama, T.; Iwashita, T.; Uemura, D. *Tetrahedron Lett.* **2007**, *48*, 3423–3427.

- Calculations were conducted using the GAUSSIAN 98 program Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Zakrzewski, V. G.; Montgomery, J. A., Jr.; Stratmann, R. E.; Burant, J. C.; Dapprich, S.; Millam, J. M.; Daniels, A. D.; Kubin, K. N.; Strain, M. C.; Farkas, O.; Tomasi, J.; Barone, V.; Cossi, M.; Cammi, R.; Mennucci, B.; Pomelli, C.; Adamo, C.; Clifford, S.; Ochterski, J.; Petersson, G. A.; Ayala, P. Y.; Cui, Q.; Morokuma, K.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Cioslowski, J.; Ortiz, J. V.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Gomperts, R.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Gonzalez, C.; Challacombe, M.; Gill, P. M.; Head-Gordon, M.; Replogle, E. S.; Pople, J. A. GAUSSIAN 98 revision A.5; Gaussian: Pittsburgh, PA, 1998.
- (a) Aho, J. E.; Pihko, P. M.; Rissa, T. K. Chem. Rev. 2005, 105, 4406; (b) Perron, F.; Albizati, K. F. Chem. Rev. 1989, 89, 1617.

4. Another stereoisomer $(17R^* \text{ and } 19R^*)$ whose A-ring was flipped was also conceivable (see below). However, the bisequatorial C–O conformation in the 6,5,6-bis-spiroacetal ring may be unfavorable from the viewpoint of anomeric effects, and the lack of NOEs regarding H17/H22 and H19/H22 disagreed with such a conformation.



- 5. Coupling constants were calculated using the electronegativity-modified Karplus equation⁶ as implemented within MacroModel.
- Hasnoot, C. A. G.; DeLeeuw, F. A. A. M.; Altona, C. *Tetrahedron* 1980, *36*, 2783.